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3 Atoms: The Building Blocks of Matter

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5 The Periodic Law - Jefferson Township Public Schools

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

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CHAPTER 10 REVIEW States of Matter
SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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CHAPTER 3 REVIEW Atoms: The Building
Blocks of Matter SECTION 3 SHORT ANSWER
Answer the following questions in the space
provided. 1. Explain the difference between the
mass number and the atomic number of a nuclide.

Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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6 Chemical Bonding

CHAPTER 5 REVIEW The Periodic Law

SECTION 3 SHORT ANSWER Answer the

following questions in the space provided. 1. c

When an electron is added to a neutral atom, energy

is (a) always absorbed. (c) either absorbed or

released. (b) always released. (d) neither absorbed

nor released. 2. d The energy required to remove an electron from a neutral ...

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7 Chemical Formulas and Chemical Compounds

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how

many sulfate ions can be attached to the iron atom.

(c) the charge on ...

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3 Atoms: The Building Blocks of Matter - Ms. Sheehan's ...



CHAPTER 9 REVIEW Stoichiometry SECTION 3

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1.

88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N_2 are mixed with 12.0 mol of

H

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CHAPTER 6 REVIEW Chemical Bonding

SECTION 4 SHORT ANSWER Answer the

following questions in the space provided. 1. b In

metals, the valence electrons are considered to be

(a) attached to particular positive ions. (c)

immobile. (b) shared by all surrounding atoms. (d)

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involved in covalent bonds.